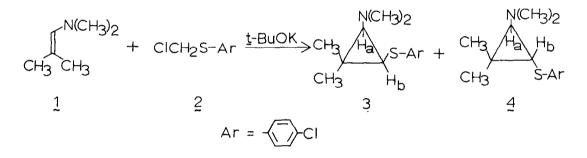
THE ADDITION OF THIOCARBENES TO ENAMINES

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There are several reported examples of the addition of carbenes (or carbenoids¹) to enamines to form cyclopropylamines.²⁻⁵ Thiocarbenes (or carbenoids), generated from chloromethyl sulfides, have been shown to add to olefins.⁶⁻⁸ This report describes the addition of thiocarbenes to enamines to form aminocyclopropyl sulfides.

Treatment of N,N,2-trimethylpropenylamine (l) (three-fold excess) with chloromethyl p-chlorophenyl sulfide (l) and t-BuOK in Et₂O at 25° resulted in the formation of a 62% yield of <u>cis</u> and trans-N,N-dimethyl-N {[1,1-dimethyl-2-(p-chlorophenyl)thio]cyclopropyl}amine (3 and 4). Adsorp-



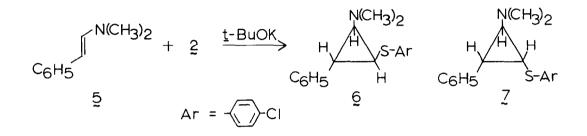
tion chromatography on silica gel afforded 3^9 , mp 59-61°, and 4, bp 100-104°/0.09 mm, in a 3.37:1.00 ratio.

Analyses of the nmr spectra show the coupling constants between H_a and H_b to be 7.0 Hz for the isomer assigned the <u>cis</u> configuration (3) and 3.5 Hz for the <u>trans</u> isomer (4). It has been established that <u>cis</u> protons on a cyclopropyl ring have larger coupling constants than do <u>trans</u> protons.^{6,8,10,11}

The observed stereoselectivity requires that the balance of substituent interactions in the transition state leads to weak net attraction. London dispersion forces, charge-transfer inter-

actions, and pure electrostatic attraction between the substituents have been invoked by Closs¹ to account for the preferred <u>syn</u> addition of substituted carbenes to <u>cis</u>-2-butene. The present results can be rationalized in terms of these interactions.

Similar treatment of β -(N,N-dimethylamino)styrene (5) with 2 afforded a mixture of <u>cis</u> and <u>trans</u>-2-[(p-chlorophenyl)thio]-N,N-dimethyl-<u>trans</u>-3-phenylcyclopropylamine (6 and 7) in 41%

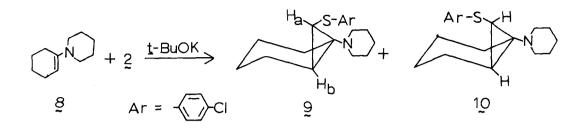


yield. Adsorption chromatography on silica gel afforded § and χ (3:2 ratio) which were subsequently converted into the hydrochlorides §g. mp 141.5-142.5, and $\chi_{g.}$ mp 160.5-161.5°.

It was not possible to assign specific nmr absorptions to individual cyclopropyl protons, and, as a result, a stereochemical assignment based on vicinyl coupling constants could not be made. A tentative stereochemical assignment based on the effect a substituent <u>cis</u> to the phenyl group in a phenylcyclopropane has on the nmr absorption of the phenyl group was made. As has been previously demonstrated,^{1,12} phenylcyclopropanes lacking a substituent <u>cis</u> to the phenyl ring show broad phenyl absorptions (relative to compounds with a <u>cis</u> substituent) because of increased shielding of the <u>ortho</u> protons. Since a significant difference in phenyl band widths was observed, the compound which possesses a broad phenyl absorption was assigned structure ξ and the compound with the sharp phenyl absorption was assigned structure 7. The above assignment is based on the assumption that a <u>trans</u> relationship between the dimethylamino and phenyl groups is maintained, i.e., the reaction is stereospecific. It has been demonstrated that the addition of phenylthiocarbene to <u>cis</u> and <u>trans</u>-2-butene is stereospecific.⁶

Similar treatment of cyclic enamine 8 with 2 and <u>t</u>-BuOK afforded a 55% yield of a mixture of 9 and 10 in <u>ca</u>. 4:1 ratio (nmr).

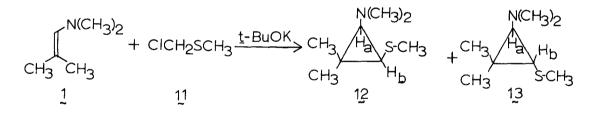
Extensive efforts to separate 2 and 10 by column chromatography were only partially successful. Small samples of the individual isomers were obtained; however, quantitative separation



could not be accomplished. Compounds 2 and 10 were converted into the hydrochlorides 2a, mp 175.5-176°, and 10a, mp 215.5-216°.

The major isomer (9) was assigned the <u>exo</u> configuration on the basis of established⁵,⁶ vicinyl coupling constants in bicyclo[4.1.0]heptanes ($J_{ab} = 5.0$ Hz). This stereochemical assignment is in accord with past stereoselectivity observations in that only a slight (1.3:1) preference for the <u>endo</u> configuration exists in the addition of phenylthiocarbene to cyclohexene;⁶ whereas, the addition of p-chlorophenylthiocarbene to enamine $\frac{1}{5}$ shows a strong preference (3.37:1) for formation of the <u>cis</u> isomer.

Application of the above reaction to chloromethyl methyl sulfide $(\frac{11}{12})$ and $\frac{1}{12}$ afforded a 15% yield of <u>cis</u> and <u>trans-N,N,2,2-tetramethyl-3-(methylthio)cyclopropylamine</u> ($\frac{12}{12}$ and $\frac{13}{12}$). Chromato-



graphic separation afforded 12; bp 65-o8°/13 mm, $J_{ab} = 7.5$ Hz, and 13, bp 58-60°/13 mm, $J_{ab} = 3.5$ Hz, in a 5:1 ratio.

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